


# Safety Data Sheet

According to Commission Regulation (EU) No 2015/830

Issue date 21/07/2016  
 Issue 3  
 Review date 27/10/2016  
 Review 4

## Soluble Ammonium Nitrate with a Nitrogen content $\geq$ 34% N

SECTION 1		Identification of the substance/mixture and of the company/undertaking			
1.1	Product identifier				
	Product commercial name	Soluble Ammonium Nitrate; FERTIBERSOL - NA; Fertibersol Antech			
	Chemical name	Ammonium nitrate			
	Other names	Soluble Ammonium Nitrate Fertilizer, Ammonium Salt and Nitric Acid, Ammonium Nitrate 34.5; Soluble Ammonium Nitrate Fertilizer (without ammonium sulphate)			
	Chemical formula	NH <sub>4</sub> NO <sub>3</sub>			
	EU index number (Appendix 1)	Not applicable			
	CE No	229-347-8			
	CAS No.	6484-52-2			
	REACH or National product registration number	01-2119490981-27-0028			
1.2	Relevant identified uses of the substance or mixture and uses advised against				
	Identified uses	Fertilizer			
	Uses advised against	Any other use.			
1.3	Details of the supplier of the safety data sheet				
	Company name	FERTIBERIA. S.A.			
	Company address	Paseo de la Castellana, 259 D. Plantas 47 y 48 - 28046 Madrid			
	Company telephone number	Central: 91.586.62.00; Puertollano factory: 926.44.93.00			
	Company email for SDS	<a href="mailto:reachfertiberia@fertiberia.es">reachfertiberia@fertiberia.es</a>			
1.4	Emergency telephone number	Puertollano factory: 926.44.93.00			
SECTION 2		Hazards identification			
2.1	Classification of the substance or mixture*	According to Regulation EC 1272/2008 [CLP] Oxidising solid, Cat3; H272 Eye Irritation, Cat2.; H319			
2.2	Label elements	Pictograms	Signal word	Hazard statements	Precautionary Statements
			Warning	H272 H319	P210 P220 P280 P305+P351+P338 P337 + P313
2.3	Other hazards				
	PBT/vPvP Criteria	In accordance with appendix XIII of the Regulation (EC) no. 1907/2006, it is not PBT or vPvB since it is an inorganic substance.			
	<b>Other hazards that do not involve product classification</b>				
	Physical and chemical hazards	Fertilizers are basically harmless products when handled correctly. Nevertheless, the following points should be noted for fire, heating and detonation: - When strongly heated it melts and if heated further it can decompose releasing toxic fumes that contain nitrogen and ammonium oxides. - These products have a high resistance to detonation. - Heating under strongly confined conditions may lead to an explosive reaction.			
	Health hazards	Fertilizers are basically harmless products when handled correctly. Nevertheless, the following points should be observed: <b>Contact with skin:</b> Prolonged contact may cause discomfort. <b>Ingestion:</b> Small quantities are unlikely to cause toxic effects. In large quantities it can produce disorders in the gastrointestinal tract and in extreme cases the formation of methemoglobin can occur (blue baby syndrome) and cyanosis (indicted by blueness around the mouth). <b>Inhalation:</b> High concentrations of dust in the air may cause nose and upper respiratory tract irritation with sore throat and cough symptoms. <b>Long term local effects:</b> Unknown adverse effects. <b>Other: Fire and heating:</b> Inhaling decomposition gases containing nitrogen and ammonium oxides can cause irritation and have corrosive effects on the respiratory system. These gases may have delayed pulmonary oedema effects.			
	Environmental hazards	Ammonium nitrate is a nitrogen fertilizer. Heavy spillage may cause an adverse environmental impact such as eutrophication (developing undesirable flora in confined surface waters or nitrate contamination. (See section 12).			
* To understand the full meaning of hazard statements (H): see section 16					

## Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

SECTION 3 Composition/information on ingredients							
3.1	Name	CE No.	CAS No.	% (w/w)	IUPAC name	Classification Regulation 1272/2008	Specific concentration limits
	Ammonium nitrate	229-347-8	6484-52-2	$\geq 98,5\%$	ammonium nitrate	Oxid. Solid 3 Eye Irrit. 2	----

May contain other substances not classified:

Less than 0,2% of combustible material expressed as "carbon" C ".

Less than 5% ammonium sulphate.

SECTION 4 First aid measures											
4.1	Description of first aid measures										
	<table border="1"> <tr> <td>General</td> <td>Seek medical attention when necessary. Inhalation of gases, from a fire or thermal decomposition, that contain nitrogen and ammonium oxides may cause irritation and have corrosive effects on the respiratory system. Administer oxygen, especially if there is blue colouring around the mouth.</td> </tr> <tr> <td>Inhalation</td> <td>Remove the person from the point of exposure to the dust. Seek medical attention if there are any harmful effects.</td> </tr> <tr> <td>Ingestion</td> <td>Do not induce vomiting. Rinse the mouth and give water or milk to drink. Seek medical attention if more than a small quantity has been ingested.</td> </tr> <tr> <td>Contact with skin</td> <td>Wash the affected area with plenty of water.</td> </tr> <tr> <td>Contact with eyes</td> <td>Wash or rinse the eyes with plenty of water for at least 15 minutes, including behind the eyelids. Remove contact lenses if present and easy to do. Seek medical attention if eye irritation persists.</td> </tr> </table>	General	Seek medical attention when necessary. Inhalation of gases, from a fire or thermal decomposition, that contain nitrogen and ammonium oxides may cause irritation and have corrosive effects on the respiratory system. Administer oxygen, especially if there is blue colouring around the mouth.	Inhalation	Remove the person from the point of exposure to the dust. Seek medical attention if there are any harmful effects.	Ingestion	Do not induce vomiting. Rinse the mouth and give water or milk to drink. Seek medical attention if more than a small quantity has been ingested.	Contact with skin	Wash the affected area with plenty of water.	Contact with eyes	Wash or rinse the eyes with plenty of water for at least 15 minutes, including behind the eyelids. Remove contact lenses if present and easy to do. Seek medical attention if eye irritation persists.
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Contact with eyes	Wash or rinse the eyes with plenty of water for at least 15 minutes, including behind the eyelids. Remove contact lenses if present and easy to do. Seek medical attention if eye irritation persists.										
4.2	Most important symptoms and effects, both acute and delayed										
	Eye irritation. Some effects on the lungs may be delayed.										
4.3	Indication of any immediate medical attention and special treatment needed										
	Inhalation of gases, from a fire or thermal decomposition, that contain nitrogen and ammonium oxides may cause irritation and have corrosive effects on the respiratory system. Administer oxygen, especially if there is blue colouring (methaemoglobin) around the mouth.										
SECTION 5 Firefighting measures											
5.1	Extinguishing media										
	<table border="1"> <tr> <td>Suitable extinguishing media</td> <td>Water.</td> </tr> <tr> <td>Unsuitable extinguishing media</td> <td>Do not use chemical or foam extinguishers or attempt to suffocate the fire with sand or mist.</td> </tr> </table>	Suitable extinguishing media	Water.	Unsuitable extinguishing media	Do not use chemical or foam extinguishers or attempt to suffocate the fire with sand or mist.						
Suitable extinguishing media	Water.										
Unsuitable extinguishing media	Do not use chemical or foam extinguishers or attempt to suffocate the fire with sand or mist.										
5.2	Special hazards arising from the substance or mixture										
	<table border="1"> <tr> <td>Special hazards</td> <td>Not combustible. If it is involved in a fire it will intensify it as it is an oxidising agent, it can maintain the fire even in the absence of air. There is a potential explosion risk during the fire when the product is strongly confined and/or contaminated with incompatible materials (e.g. organic material, halogen compounds - see section 10) Prilled product must not be put in drains.</td> </tr> <tr> <td>Thermal decomposition or product combustion hazards</td> <td>Nitrogen and ammonium oxides</td> </tr> </table>	Special hazards	Not combustible. If it is involved in a fire it will intensify it as it is an oxidising agent, it can maintain the fire even in the absence of air. There is a potential explosion risk during the fire when the product is strongly confined and/or contaminated with incompatible materials (e.g. organic material, halogen compounds - see section 10) Prilled product must not be put in drains.	Thermal decomposition or product combustion hazards	Nitrogen and ammonium oxides						
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Thermal decomposition or product combustion hazards	Nitrogen and ammonium oxides										
5.3	Advice for firefighters										
	<table border="1"> <tr> <td>Specific firefighting methods</td> <td>Open doors and windows in the area to give maximum ventilation. Avoid breathing the smoke (toxic). Position yourself upwind of the fire. Prevent contamination of the fertiliser by oils or other combustible materials.</td> </tr> <tr> <td>Special protective equipment for firefighting</td> <td>Use self contained breathing apparatus in case of smoke.</td> </tr> </table>	Specific firefighting methods	Open doors and windows in the area to give maximum ventilation. Avoid breathing the smoke (toxic). Position yourself upwind of the fire. Prevent contamination of the fertiliser by oils or other combustible materials.	Special protective equipment for firefighting	Use self contained breathing apparatus in case of smoke.						
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Special protective equipment for firefighting	Use self contained breathing apparatus in case of smoke.										
SECTION 6 Accidental release measures											
6.1	Personal precautions, protective equipment and emergency procedures										
	Avoid walking on the spilt product and exposure to the dust.										
6.2	Environmental precautions										
	Take care to prevent contamination of water courses and drains and inform the competent authorities in case of accidental contamination of water courses.										
6.3	Methods and material for containment and cleaning up										
	Any spillage of fertiliser should be quickly cleaned up, swept and placed in a clean, labelled, open receptacle for safe disposal. Do not mix with sawdust or other combustible or organic material. Dilute any contaminated or fine grain product with inert materials such as limestone/dolomite, mineral phosphate, gypsum, sand or dissolve in water.										
6.4	Reference to other sections										
	See section 8 for personal protective equipment and section 13 for the disposal of waste										

## Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

SECTION 7		Handling and storage
7.1	Precautions for safe handling	
		Prevent the excessive generation of dust. Prevent contamination with combustible materials (e.g. gas-oil, greases, etc.) and other incompatible materials. Avoid the unnecessary exposure of the product to the atmosphere to prevent moisture absorption. When the product is handled for long periods, use appropriate personal protective equipment, e.g. gloves. Carefully clean the installations before carrying out maintenance and repair operations.
7.2	Conditions for safe storage, including any incompatibilities	
		Store in compliance with ITC-MIE-APQ08 regulations. Place away from sources of heat and flames. Always keep away from combustible materials and substances mentioned in section 10. In the field, ensure that the product is not stored near hay, straw, grain, gas-oil, etc. In the storage area, ensure that strict tidiness and cleanliness standards are complied with. Do not allow smoking or the use of naked portable lamps in the storage area. Restrict the size of piles and stacks (in accordance with local regulations) and leave a minimum free space of 1 metre around the piles or stacks of sacks. Any building used for storage should be clean and well ventilated. When required, due to the nature of the product in containers and weather conditions, the product should be stored in such a way as to avoid its destruction due to thermal cycles (extreme temperature conditions). The product should not be stored in direct sunlight to prevent physical break-up due to thermal cycles.
	Recommended and non-recommended packaging materials	Suitable materials for containers are: steel, aluminium and synthetic plastics. Do not use copper and/or zinc.
7.3	Specific end use(s)	
		See section 1.2 and appendices for exposure scenarios.
<i>Note: stability and reactivity, see section 10</i>		

SECTION 8		Exposure controls/personal protection						
8.1	Control parameters							
	Exposure limit values		Component	CAS				
			Ammonium nitrate	6484-52-2	Not established.			
	Derived from the CSR	DNEL			Worker			consumer
				systemic	industrial	professional		
			oral	long term	Not applicable	Not applicable	12.8 mg/kg bw/day	
			inhalation	long term	37.6 mg/m <sup>3</sup>	37.6 mg/m <sup>3</sup>	11.1 mg/m <sup>3</sup>	
			dermal	long term	21.3 mg/Kg bw/day	21.3 mg/Kg bw/day	12.8 mg/kg bw/day	
	PNEC	water	air	soil	microbiological	sediment	oral	
		fresh water: 0.45 mg/l salt water: 0.045 mg/l in intermittent releases: 4.5 mg/l	Not available	Insufficient data available	18 mg/l	Insufficient data available	Low bioaccumulative potential	
8.2	Exposure controls							
	Engineering measures and hygiene controls		Prevent high concentrations of dust and provide ventilation wherever necessary. Do not smoke or drink when handling. Wash hands after handling the product and before eating, drinking or smoking. Use the wash basin at the end of the work day.					
	Personal protection measures							
		Eyes	Safety glasses with side protection (EN 166) to prevent eye irritation. In dusty conditions use panoramic safety goggles.					
		Skin and body	Work clothes.					
		Hands	Use suitable gloves (for example, rubber or leather) when handling the product over long periods of time.					
		Respiratory	If there is a high concentration of dust and/or the ventilation is inadequate, use an anti-dust mask or respirator with a suitable filter.					
		Thermal						
	Environmental exposure controls		See section 6.					
								<i>Advice relating to personal protection is valid for high exposure levels.</i>
								<i>Choose personal protection equipment suitable to exposure risks.</i>

## Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

SECTION 9		Physical and chemical properties																	
9.1		Information on basic physical and chemical properties																	
Aspect		White or coloured granules or prills.																	
Colour		White or coloured																	
Odour		Odourless																	
Molecular weight		80																	
pH		pH aqueous solution (100 g/l) > 4.5.																	
Boiling point		It does not have a boiling point, it decomposes above 210°C																	
Melting point		169 °C																	
Flash-point		Non flammable																	
Flammability		Non flammable																	
Explosive properties		<p><b>Ammonium nitrate with less than 0.2% combustible material (UN 1942) is not classified as an explosive.</b>                      Fertiliser grade ammonium nitrate (UN 2067 or UN 2071) does not have explosive properties.                      Liquid ammonium nitrate (UN2426) is not classified as an explosive.                      Ammonium nitrate with &gt; 0.2% of combustible material (UN 0222) is classified as an explosive material (Class 1).                      The UN series 1 and 2 tests show that crystalline ammonium nitrate with no impurities is not a class 1 explosive material.</p> <p>If it is heated under strongly confined conditions (e.g. in pipes or drains) a violent reaction or explosion may take place, especially if there is contamination by any of the substances mentioned in section 10.</p>																	
Auto-ignition temperature		Non flammable																	
Decomposition temperature		Begins to decompose above 170 °C																	
Lower explosive limit		Not applicable																	
Upper explosive limit		Not applicable																	
Oxidising properties		<p><b>UN1942</b> and UN2067: <b>Classification for transport: Class 5.1; GE III.</b>                      UN2426: Classification for transport: Class 5.1; GE unknown.                      UN0222: Classification for transport: Class 1.1D (explosive).                      Fertilisers with UN2071 are non oxidising.</p>																	
Apparent density at 20°C		650 at 800 kg/m <sup>3</sup>																	
Vapour pressure at 20 °C		Not applicable																	
Vapour density		Not applicable																	
Partition coefficient n-octanol/water		Not applicable																	
Viscosity		Not applicable																	
Water solubility		> 100 g/l (hygroscopic)																	
9.2		Additional information																	
		No relevant data																	
SECTION 10		Stability and reactivity																	
10,1		Reactivity																	
		Stable under normal conditions of storage, handling and use																	
10,2		Chemical stability																	
		Stable under normal conditions of storage, handling and use																	
10,3		Possibility of hazardous reactions																	
		When it is heated above 170°C it decomposes releasing NOx and ammonia. Contamination with incompatible materials.																	
10,4		Conditions to avoid																	
		Proximity to sources of heat or fire. Contamination by incompatible materials. Unnecessary exposure to the atmosphere. Heating when confined. Welding or heating work of the equipment or plant that may contain product remnants, without preliminary cleaning to remove the product remnants.																	
10,5		Incompatible materials																	
		Inflammable materials, reducing agents, acids, alkalis, sulphur, chlorates, chlorides, chromates, nitrites, permanganate, metal powders and metal-containing substances such as copper, nickel, cobalt, zinc and their alloys.																	
10,6		Hazardous decomposition products																	
		In case of fire: see Section 5 When strongly heated it melts and decomposes releasing toxic gases (e.g. NOx and ammonia). When it is in contact with alkaline materials, such as lime, ammonia gases may be produced.																	
SECTION 11		Toxicological information																	
11,1		Information on toxicological effects																	
		Acute toxicity																	
		<table border="1"> <thead> <tr> <th>Component</th> <th>CAS No.</th> <th>Method</th> <th>Species</th> <th>Via</th> <th>Result</th> </tr> </thead> <tbody> <tr> <td>Ammonium nitrate</td> <td>6484-52-2</td> <td>OECD 401 OECD 402</td> <td>rat rat rat</td> <td>oral skin respiratory</td> <td>LD50: 2950 mg/Kg bw. LD50: &gt; 5000 mg/Kg bw. LC50: &gt;88.8 mg/m3.</td> </tr> </tbody> </table>						Component	CAS No.	Method	Species	Via	Result	Ammonium nitrate	6484-52-2	OECD 401 OECD 402	rat rat rat	oral skin respiratory	LD50: 2950 mg/Kg bw. LD50: > 5000 mg/Kg bw. LC50: >88.8 mg/m3.
Component	CAS No.	Method	Species	Via	Result														
Ammonium nitrate	6484-52-2	OECD 401 OECD 402	rat rat rat	oral skin respiratory	LD50: 2950 mg/Kg bw. LD50: > 5000 mg/Kg bw. LC50: >88.8 mg/m3.														
		Skin corrosion/irritation																	
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Component	CAS No.	Method	Species	Via	Result														
Ammonium nitrate	6484-52-2	OECD 404	Rabbit	skin	Non-irritant.														
		Serious eye damage/irritation																	
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Component	CAS No.	Method	Species	Via	Result														
Ammonium nitrate	6484-52-2	OECD 405	Rabbit	eye	Irritant														
		Respiratory or skin sensitisation																	
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Component	CAS No.	Method	Species	Via	Result														
Ammonium nitrate	6484-52-2	OECD 429	mouse	skin	Non-sensitising.														
		Germ cell mutagenicity																	
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Component	CAS No.	Method	Species	Via	Result														
Ammonium nitrate	6484-52-2	OECD 471 OECD 473 OECD 476	bacteria Chromosomal aberrations mutation in mammal cells		Negative. Non-mutagenic. Ames test. Negative. Non-mutagenic. Negative. Non-mutagenic.														

## Soluble Ammonium Nitrate with a Nitrogen content $\geq$ 34% N

Carcinogenicity					
Component	CAS No.	Method	Species	Via	Result
Ammonium nitrate	6484-52-2		rat	All	Non carcinogenic.
Reproductive toxicity					
Component	CAS No.	Method	Species	Via	Result
Ammonium nitrate	6484-52-2	OECD 422	rat	oral	-Effects on fertility: NOAEL: $\geq$ 1500 mg/kg bw/d. -Toxicity for development: NOAEL: $\geq$ 1500 mg/kg bw/d
STOT-single exposure					
Component	CAS No.	Method	Species	Via	Result
Ammonium nitrate	6484-52-2	-	-	-	Not available
STOT-repeated exposure					
Component	CAS No.	Method	Species	Via	Result
Ammonium nitrate	6484-52-2	OECD 422 OECD 453	rat rat	oral (28 days) oral (52 weeks)	Sub-acute oral route. NOAEL: $\geq$ 1500 mg/kg body weight/day. Chronic oral route. NOAEL: 256 mg/kg body weight/day.
Aspiration hazards					
Component	CAS No.	Method	Species	Via	Result
Ammonium nitrate	6484-52-2	-	-	-	See note below
Inhaling high concentrations of dust may cause nose and upper respiratory tract irritation with sore throat and cough symptoms.					

### SECTION 12 Ecological information

12,1	Toxicity					
Water toxicity						
Component	CAS No.		Fish (Cyprinus carpio)	Crustaceans	Algae (benthic diatoms)	
Ammonium nitrate	6484-52-2	Short term	LC50(48h) = 447 mg/l.	EC50/LC50 (48h) = 490 mg/l (of potassium nitrate) (Daphnia magna)	LC50/EC50 (10 days) > 1700 mg/l (of potassium nitrate)	
		Long term	Not necessary.	NOEC (168h) = 555 mg/l (Bullia digitalis)	Not available	
Land Toxicity						
Component	CAS No.	Macroorganisms	Microorganisms	Land plants	Other organisms	
Ammonium nitrate	6484-52-2	Not scientifically justified	Not scientifically justified	Not scientifically justified	Not available	
Microbiological activity in waste water treatment plants						
Component	CAS No.	Toxicity for aquatic microorganisms				
Ammonium nitrate	6484-52-2	CE50/CL50 (180 min) >1000 mg/l (of sodium nitrate)				
12,2	Persistence and degradability					
Component	CAS No.	Degradation				
Ammonium nitrate	6484-52-2	Hydrolysis	Non-hydrolysable. Test not necessary.			
		Photolysis	No information available			
		Biodegradation	Not necessary, inorganic substance.			
12,3	Bioaccumulative potential					
Component	CAS No.	Octanol-water partition coefficient (Kow)	Bioconcentration factor (BCF)	Comments		
Ammonium nitrate	6484-52-2	Not applicable. Inorganic substance.	-			
12,4	Mobility in soil					
Component	CAS No.	Result				
Ammonium nitrate	6484-52-2	low absorption potential (based on its properties)				
12,5	Results of PBT and vPvB assessment					
Not required. Inorganic substance. See REACH appendix XIII.						
12,6	Other adverse effects					
Heavy spillage may cause an adverse environmental impact such as eutrophication in confined surface waters.						


### SECTION 13 Disposal considerations

13,1	Waste treatment methods					
Depending on the degree and nature of the contamination, dispose of it as a fertiliser over the ground or in an authorised waste installation. Do not put the waste in the drain, dispose of the product waste and containers in a safe way. Dispose of in accordance with all local and national regulations. Empty containers by shaking them to remove as much as possible of their content. If approved by the local authorities, empty packaging can be disposed of as a non-hazardous material or returned for recycling.						

### SECTION 14 Transport Information

14.1 - 14.6	Regulatory Information	UN Number	Proper shipping name	Class	Packing group	Label	Environmental hazards	Special precautions for users

## Soluble Ammonium Nitrate with a Nitrogen content $\geq$ 34% N

ADR/RID ADNR IMDG IATA	UN 2067	AMMONIUM NITRATE BASED FERTILISERS	5,1	III		NO	Hazard identification number: 50 See ADR and RID
							Emergency procedures (EmS): F-H, S-Q
							See ICAO regulation for quantity limitation
	<b>14,7</b> <i>Transport in bulk according to Annex II of Marpol and the IBC Code: Not applicable</i>						

### SECTION 15 Regulatory information

<b>15,1</b>	<b>Safety, health and environmental regulations/legislation specific for the substance or mixture</b>
	<p>Regulation 2003/2003 (fertilisers)                  Regulation 1907/2006 (REACH). Entry 58 of appendix XVII.                  Regulation 1272/2008 (CLP)                  Directive 18/2012 (Seveso Directive)                  R.D. 840/2015 (Seveso)                  R.D. 506/2013 (fertilizers)                  R.D. 374/2001 (Chemical agents)                  R.D. 145/1989: National Regulation for Admission, Handling and Storage of Hazardous Materials in ports.                  RD. 2492/1983 of 29 June. State Administrative Intervention on explosive grade Ammonium Nitrates.                  RD. 2016/2004: ITS MIE APQ-8 Storage of ammonium nitrate based fertilisers with a high nitrogen content.                  ORDER PRE. 988/2004: Detonability test for ammonium nitrate based products with a high nitrogen content.</p>
<b>15,2</b>	<b>Chemical Safety Assessment</b>
	Chemical Safety Assessment for ammonium nitrate

### SECTION 16 Other information

<b>Hazard statements</b>	H272: May intensify a fire; oxidizer. H319: Causes serious eye irritation.
<b>Precautionary statements</b>	P102: Keep out of reach of children. P210: Keep away from heat, sparks, open flames and hot surfaces. No smoking. P220: Keep/Store away from clothing/combustible materials. P264: Wash hands thoroughly after handling. P280: Wear eye protection. P305+P351+P338: IF IN EYES: Rinse continuously with water for several minutes. Remove contact lenses if present and easy to do. Continue rinsing. P337+P313: If eye irritation persists: get medical attention. P370+P378: In case of fire: Use water for extinction.
<b>Bibliographical references and data sources</b>	Ammonium nitrate chemical safety assessment. Guidance documents EFMA/FERTILIZER EUROPE; Data for TFI HPV; NOTOX
<b>Abbreviations and acronyms</b>	ELV-DE: Environmental limit value (daily exposure) ELV-ST Environmental limit value (short term) NOAEL: No observable adverse effect level LD50: Lethal dose 50% LC50: Lethal concentration 50% EC50: Effective concentration 50% DNEL: Derived no effect level PNEC: Predicted no effect concentration LOEC: Lowest observed effect concentration NOEC: No observed effect concentration NOAEC: No observed adverse effect concentration
<b>Adequate training for workers</b>	Obligatory training in occupational risk prevention
<b>Date of prior SDS</b>	Version 3 dated 21.07.16
<b>Modifications made to present revision</b>	Incorporation of the Fertibersol Antech

**Exposure scenarios 1,2,3 and 4 are attached**

The information contained in this Safety Data Sheet is given in good faith. It is accurate to the best of our knowledge and belief and represents the most up to date information about the product at the time of publication. The information given in this data sheet does not constitute or replace the user's own assessment of workplace risks as required by other health and safety legislation.

# Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

## Safety Data Sheet Appendices Exposure Scenario 1

<b>1</b>	<b>Title of Exposure Scenario (ES)</b>																		
	Manufacture of ammonium nitrate																		
<b>2</b>	<b>Description of activities or processes covered by the exposure scenario</b>																		
	<p><b>List of all the use descriptors related to ES 1</b></p> <p>SU 8/9 * PROC 1/2/3/8a/8b/9/14/15 ERC 1</p> <p><b>Name/s of contributing scenario/s related to the environment and their corresponding Environmental Release Class (ERC)</b></p> <p>1. Manufacture of substances (ERC 1)</p> <p><b>Name/s of contributing scenario/s for the worker and their corresponding Process Category (PROC)</b></p> <p>1. Use in enclosed processes, no likelihood of exposure (PROC 1) 2. Use in closed, continuous processes with occasional controlled exposure (PROC 2) 3. Use in closed batch processes (synthesis or formulation) (PROC 3) 4. Transfer of substances or preparations (charging/discharging) from/to vessels/large containers at non-dedicated facilities (PROC 8a) 5. Transfer of substances or preparations (charging/discharging) from/to vessels/large containers at dedicated facilities (PROC 8b) 6. Transfer of substances or preparations into small containers (dedicated filling line, including weighing) (PROC 9) 7. Production of mixtures or articles by tableting, compression, extrusion, pelletisation (PROC 14) 8. Use as laboratory reagent (PROC 15)</p> <p>* Agency Guidance Document, Chapter R.12: Use descriptor system: SU8 (Manufacture of bulk, large scale chemicals)/ SU9 (Manufacture of fine chemicals)</p>																		
<b>2.1</b>	<b>Contributing scenario (1) controlling environmental exposure for the manufacture of ammonium nitrate (ES1)</b>																		
	<p>Environmental exposure due to the manufacture of ammonium nitrate</p> <p>Section 2.1 describes emissions to the environment that can occur during the manufacture of ammonium nitrate (ERC 1).</p> <p>As this substance does fulfil criteria for classification as hazardous to the environment, the environmental risk assessment has not been carried out for this substance and therefore the conditions that affect the environment are not included during this use.</p>																		
<b>2.2</b>	<b>Contributing scenario (2) controlling exposure of workers for manufacture of the substance, including handling, storage and quality controls</b>																		
	<p>Section 2.2 describes potential exposure of workers from manufacture of the substance, including handling, storage and quality controls.</p> <p>All the relevant processes for the contributing scenarios identified by the PROC codes in point 1 of this scenario (PROC 1/2/3/8a/8b/9/14/15) have the same operating conditions and risk management measures for personnel. Consequently they are all covered by just one contributing scenario (2).</p>																		
	<table border="1"> <tr> <td><b>Product characteristics</b></td> <td>Solid with low dust formation index</td> </tr> <tr> <td><b>Quantities used</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Frequency and duration of use or exposure</b></td> <td>&gt; 4 hours a day</td> </tr> <tr> <td><b>Human factors not influenced by risk management</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Other operational conditions that have an impact on worker exposure</b></td> <td>Ammonium nitrate is manufactured in enclosed environments.</td> </tr> <tr> <td><b>Technical conditions and measures at process level (source) to prevent release</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Technical conditions and measures for controlling dispersion of the source to workers</b></td> <td>1.- Adequate containment of the substance 2.- Good ventilation conditions</td> </tr> <tr> <td><b>Organisational measures to prevent or limit releases, dispersion and exposure</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Conditions and measures for personal protection, hygiene and health evaluation</b></td> <td>Use safety glasses</td> </tr> </table>	<b>Product characteristics</b>	Solid with low dust formation index	<b>Quantities used</b>	Not relevant	<b>Frequency and duration of use or exposure</b>	> 4 hours a day	<b>Human factors not influenced by risk management</b>	Not relevant	<b>Other operational conditions that have an impact on worker exposure</b>	Ammonium nitrate is manufactured in enclosed environments.	<b>Technical conditions and measures at process level (source) to prevent release</b>	Not relevant	<b>Technical conditions and measures for controlling dispersion of the source to workers</b>	1.- Adequate containment of the substance 2.- Good ventilation conditions	<b>Organisational measures to prevent or limit releases, dispersion and exposure</b>	Not relevant	<b>Conditions and measures for personal protection, hygiene and health evaluation</b>	Use safety glasses
<b>Product characteristics</b>	Solid with low dust formation index																		
<b>Quantities used</b>	Not relevant																		
<b>Frequency and duration of use or exposure</b>	> 4 hours a day																		
<b>Human factors not influenced by risk management</b>	Not relevant																		
<b>Other operational conditions that have an impact on worker exposure</b>	Ammonium nitrate is manufactured in enclosed environments.																		
<b>Technical conditions and measures at process level (source) to prevent release</b>	Not relevant																		
<b>Technical conditions and measures for controlling dispersion of the source to workers</b>	1.- Adequate containment of the substance 2.- Good ventilation conditions																		
<b>Organisational measures to prevent or limit releases, dispersion and exposure</b>	Not relevant																		
<b>Conditions and measures for personal protection, hygiene and health evaluation</b>	Use safety glasses																		
<b>3</b>	<b>Estimation of exposure and reference to its source</b>																		
	<p><b>Information for contributing scenario 1 (environmental exposure):</b></p> <p>No environmental assessment was completed because the substance does not fulfil criteria for classification as hazardous for the environment and therefore there is no additional assessment of environmental exposure.</p> <p><b>Information for contributing scenario 2 (exposure for personnel):</b></p> <p>A qualitative assessment has been included that concludes that this use is safe for workers.</p> <p>The toxicological effect of this substance is eye irritation (local parameter), for which a DNEL value cannot be estimated, because there is no dose-response information available. A minimal systemic effect was only observed at doses that were so high that personnel would never be exposed to them (see relevant DNEL: section 8 SDS), so it was not considered necessary to make a quantitative risk assessment.</p>																		
<b>4</b>	<b>Guidance for intermediate users to assess if they working within the limits set by the ES</b>																		
	Additional risk management measures apart from the ones mentioned above in the contributing scenarios (2.1, 2.2) are not required to guarantee safety during this use and thus work within the limits of the ES 1 exposure scenario.																		
<b>5</b>	<b>Good practice advice in addition to that included in the Chemical Safety Assessment (CSA) required by REACH. Measures not subject to art. 37 (4) REACH</b>																		
	<ul style="list-style-type: none"> <li>- Adequate containment of the substance</li> <li>- Minimise the number of exposed personnel</li> <li>- Segregate the releasing processes</li> <li>- Utilise effective contamination extraction systems</li> <li>- Good ventilation conditions</li> <li>- Minimise manual handling</li> <li>- Avoid contact with contaminated objects and instruments</li> <li>- Regularly clean the work area and equipment</li> <li>- Supervise the area to check that risk management measures are being applied</li> <li>- Train personnel for good practices</li> <li>- Keep standard personal hygiene conditions</li> </ul>																		

# Soluble Ammonium Nitrate with a Nitrogen content $\geq$ 34% N

## Safety Data Sheet Appendices Exposure Scenario 2

<b>1</b>	<b>Title of Exposure Scenario (ES)</b>	Industrial use of ammonium nitrate for formulating mixtures/articles, as intermediate substance and for end use by industry
<b>2</b>	<b>Description of activities or processes covered by the exposure scenario</b>	<p><b>List of all the use descriptors related to ES 1</b></p> <p>SU 3/10 * PC 1/11/12/19/37 * PROC 1/2/3/5/8a/8b/9/13/15 ERC 2/6a</p> <p><b>Name/s of contributing scenario/s related to the environment and their corresponding Environmental Release Class (ERC)</b></p> <p>1. Formulation of preparations (ERC 2) 2. Industrial use resulting in manufacture of another substance (use of intermediates) (ERC 6a)</p> <p><b>Name/s of contributing scenario/s for the worker and their corresponding Process Category (PROC)</b></p> <p>1. Use in enclosed processes, no likelihood of exposure (PROC 1) 2. Use in closed, continuous processes with occasional controlled exposure (PROC 2) 3. Use in closed batch processes (synthesis or formulation) (PROC 3) 4. Mixing or blending in batch processes (multistage and/or significant contact) (PROC 5) 5. Transfer of substances or preparations (charging/discharging) from/to vessels/large containers at non-dedicated facilities (PROC 8a) 6. Transfer of substances or preparations (charging/discharging) from/to vessels/large containers at dedicated facilities (PROC 8b) 7. Transfer of substances or preparations into small containers (dedicated filling line, including weighing) (PROC 9) 8. Treatment of articles by dipping and pouring (PROC 13) 9. Use as laboratory reagent (PROC 15)</p> <p>* Agency Guidance Document, Chapter R.12: Use descriptor systems: SU 3 (Industrial manufacturing: Use of substances as such or in preparations in industrial facilities) / SU10 (Formulation of preparations and/or repackaging (excluding alloys). PC 1(Adhesives, sealants)/ 11 (Explosives) /12 (Fertilisers) /19 (Intermediate substances) /37 (Chemical products for water treatment)</p>
<b>2.1</b>	<b>Contributing scenario (1) controlling environmental exposure for formulating preparations and industrial use as intermediate substance (ES 2)</b>	<p>Environmental exposure due to formulation of preparations and industrial use of ammonium nitrate as intermediate substance</p> <p>Section 2.1 describes emissions to the environment that can occur during the formulation of preparations (ERC 2) and industrial use as an intermediate substance (ERC 6a).</p> <p>As this substance does fulfil criteria for classification as hazardous to the environment, the environmental risk assessment has not been carried out for this substance and therefore the conditions that affect the environment are not included during this use.</p>
<b>2.2</b>	<b>Contributing scenario (2) controlling exposure of workers that corresponds to the industrial use of ammonium nitrate for formulating preparations/articles, as intermediate substance and end use by workers in industrial facilities</b>	<p>All the relevant processes for this scenario identified by the PROC codes in point 1 of this scenario (PROC 1/2/3/5/8a/8b/9/13/15) have the same operating conditions and risk management measures for personnel. Consequently they are all covered by just one contributing scenario (2).</p>
	<b>Product characteristics</b>	Solid with low dust formation index Liquid
	<b>Quantities used</b>	Not relevant
	<b>Frequency and duration of use or exposure</b>	> 4 hours a day
	<b>Human factors not influenced by risk management</b>	Not relevant
	<b>Other operational conditions that have an impact on worker exposure</b>	Used in enclosed spaces
	<b>Technical conditions and measures at process level (source) to prevent release</b>	Not relevant
	<b>Technical conditions and measures for controlling dispersion of the source to workers</b>	1.- Adequate containment of the substance 2.- Good ventilation conditions
	<b>Organisational measures to prevent or limit releases, dispersion and exposure</b>	Not relevant
	<b>Conditions and measures for personal protection, hygiene and health evaluation</b>	Use safety glasses
<b>3</b>	<b>Estimation of exposure and reference to its source</b>	<p><b>Information for contributing scenario 1 (environmental exposure):</b></p> <p>No environmental assessment was completed because the substance does not fulfil criteria for classification as hazardous for the environment and therefore there is no additional assessment of environmental exposure.</p> <p><b>Information for contributing scenario 2 (exposure for personnel):</b></p> <p>A qualitative assessment has been included that concludes that this use is safe for workers.</p> <p>The toxicological effect of this substance is eye irritation (local parameter), for which a DNEL value cannot be estimated, because there is no dose-response information available. A minimal systemic effect was only observed at doses that were so high that personnel would never be exposed to them (see relevant DNEL: section 8 SDS), so it was not considered necessary to make a quantitative risk assessment.</p>
<b>4</b>	<b>Guidance for intermediate users to assess if they working within the limits set by the ES</b>	Additional risk management measures apart from the ones mentioned above in the contributing scenarios (2.1, 2.2) are not required to guarantee safety during this use and thus work within the limits of the ES 2 exposure scenario.
<b>5</b>	<b>Good practice advice in addition to that included in the Chemical Safety Assessment (CSA) required by REACH. Measures not subject to art. 37 (4) REACH</b>	<ul style="list-style-type: none"> <li>- Adequate containment of the substance</li> <li>- Minimise the number of exposed personnel</li> <li>- Segregate the releasing processes</li> <li>- Utilise effective contamination extraction systems</li> <li>- Good ventilation conditions</li> <li>- Minimise manual handling</li> <li>- Avoid contact with contaminated objects and instruments</li> <li>- Regularly clean the work area and equipment</li> <li>- Supervise the area to check that risk management measures are being applied</li> <li>- Train personnel for good practices</li> <li>- Keep standard personal hygiene conditions</li> </ul>



# Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

## Safety Data Sheet Appendices Exposure Scenario 3

<b>1</b>	<b>Title of Exposure Scenario (ES)</b>																		
	Professional use of ammonium nitrate for formulation of preparations and end use by professionals																		
<b>2</b>	<b>Description of activities or processes covered by the exposure scenario</b>																		
	<p><b>List of all the use descriptors related to ES 3</b></p> <p>SU 22 * PC 12 * PROC 1/2/8a/8b/9/11/15/19 ERC 8b/8e</p> <p><b>Name/s of contributing scenario/s related to the environment and their corresponding Environmental Release Class (ERC)</b></p> <p>1. Wide dispersive indoor use of reactive substances in open systems (ERC 8b) 2. Wide dispersive outdoor use of reactive substances in open systems (ERC 8e)</p> <p><b>Name/s of contributing scenario/s for the worker and their corresponding Process Category (PROC)</b></p> <p>1. Use in enclosed processes, no likelihood of exposure (PROC 1) 2. Use in closed, continuous processes with occasional controlled exposure (PROC 2) 3. Transfer of substances or preparations (charging/discharging) from/to vessels/large containers at non-dedicated facilities (PROC 8a) 4. Transfer of substances or preparations from/to vessels/large containers at dedicated facilities (PROC 8b) 5. Transfer of substances or preparations into small containers (dedicated filling line, including weighing) (PROC 9) 6. Non industrial spraying (PROC 11) 7. Use as laboratory reagent (PROC 15) 8. Hand mixing with intimate contact (only PPE available) (PROC 19)</p> <p>* Agency Guidance Document, Chapter R.12: Use descriptor systems: SU 22 (Professional uses: Public domain (administration, education, entertainment, services, craftsmen) PC 12 (Fertilisers)</p>																		
<b>2.1</b>	<b>Contributing scenario (1) controlling environmental exposure for use of ammonium nitrate by professionals (ES3)</b>																		
	<p>Environmental exposure due to use of ammonium nitrate by professionals</p> <p>Section 2.1 describes the emissions to the environment that may occur during wide dispersive indoor use of reactive substances in open systems (ERC 8b) and wide dispersive outdoor use of reactive substances in open systems (ERC 8e)</p> <p>As this substance does fulfil criteria for classification as hazardous to the environment, the environmental risk assessment has not been carried out for this substance and therefore the conditions that affect the environment are not included during this use.</p>																		
<b>2.2</b>	<b>Contributing scenario (2) controlling worker exposure for professional use of ammonium nitrate for formulating preparations and end use.</b>																		
	All the relevant processes for this scenario identified by the PROC codes in point 1 of this scenario (PROC 1/2/8a/8b/9/11/15/19) have the same operating conditions and risk management measures for personnel. Consequently they are all covered by just one contributing scenario (2).																		
	<table border="1"> <tr> <td><b>Product characteristics</b></td> <td>Solid with low dust formation index Liquid, ammonium nitrate concentration in the product &gt; 25%</td> </tr> <tr> <td><b>Quantities used</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Frequency and duration of use or exposure</b></td> <td>&gt; 4 hours a day</td> </tr> <tr> <td><b>Human factors not influenced by risk management</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Other operational conditions that have an impact on worker exposure</b></td> <td>Used indoors and outdoors</td> </tr> <tr> <td><b>Technical conditions and measures at process level (source) to prevent release</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Technical conditions and measures for controlling dispersion of the source to workers</b></td> <td>1.- Adequate containment of the substance 2.- Good ventilation conditions 3.- Avoid splashing. Use specific dispensers and pumps designed especially to prevent splashes/leaks/exposure</td> </tr> <tr> <td><b>Organisational measures to prevent or limit releases, dispersion and exposure</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Conditions and measures for personal protection, hygiene and health evaluation</b></td> <td>Use safety glasses</td> </tr> </table>	<b>Product characteristics</b>	Solid with low dust formation index Liquid, ammonium nitrate concentration in the product > 25%	<b>Quantities used</b>	Not relevant	<b>Frequency and duration of use or exposure</b>	> 4 hours a day	<b>Human factors not influenced by risk management</b>	Not relevant	<b>Other operational conditions that have an impact on worker exposure</b>	Used indoors and outdoors	<b>Technical conditions and measures at process level (source) to prevent release</b>	Not relevant	<b>Technical conditions and measures for controlling dispersion of the source to workers</b>	1.- Adequate containment of the substance 2.- Good ventilation conditions 3.- Avoid splashing. Use specific dispensers and pumps designed especially to prevent splashes/leaks/exposure	<b>Organisational measures to prevent or limit releases, dispersion and exposure</b>	Not relevant	<b>Conditions and measures for personal protection, hygiene and health evaluation</b>	Use safety glasses
<b>Product characteristics</b>	Solid with low dust formation index Liquid, ammonium nitrate concentration in the product > 25%																		
<b>Quantities used</b>	Not relevant																		
<b>Frequency and duration of use or exposure</b>	> 4 hours a day																		
<b>Human factors not influenced by risk management</b>	Not relevant																		
<b>Other operational conditions that have an impact on worker exposure</b>	Used indoors and outdoors																		
<b>Technical conditions and measures at process level (source) to prevent release</b>	Not relevant																		
<b>Technical conditions and measures for controlling dispersion of the source to workers</b>	1.- Adequate containment of the substance 2.- Good ventilation conditions 3.- Avoid splashing. Use specific dispensers and pumps designed especially to prevent splashes/leaks/exposure																		
<b>Organisational measures to prevent or limit releases, dispersion and exposure</b>	Not relevant																		
<b>Conditions and measures for personal protection, hygiene and health evaluation</b>	Use safety glasses																		
<b>3</b>	<b>Estimation of exposure and reference to its source</b>																		
	<p><b>Information for contributing scenario 1 (environmental exposure):</b></p> <p>No environmental assessment was completed because the substance does not fulfil criteria for classification as hazardous for the environment and therefore there is no additional assessment of environmental exposure.</p> <p><b>Information for contributing scenario 2 (exposure for personnel):</b></p> <p>A qualitative assessment has been included that concludes that this use is safe for workers.</p> <p>The toxicological effect of this substance is eye irritation (local parameter), for which a DNEL value cannot be estimated, because there is no dose-response information available. A minimal systemic effect was only observed at doses that were so high that personnel would never be exposed to them (see relevant DNEL: section 8 SDS), so it was not considered necessary to make a quantitative risk assessment.</p>																		
<b>4</b>	<b>Guidance for intermediate users to assess if they working within the limits set by the ES</b>																		
	Additional risk management measures apart from the ones mentioned above in the contributing scenarios (2.1, 2.2) are not required to guarantee safety during this use and thus work within the limits of the ES 3 exposure scenario.																		
<b>5</b>	<b>Good practice advice in addition to that included in the Chemical Safety Assessment (CSA) required by REACH. Measures not subject to art. 37 (4) REACH</b>																		
	<ul style="list-style-type: none"> <li>- Adequate containment of the substance</li> <li>- Minimise the number of exposed personnel</li> <li>- Segregate the releasing processes</li> <li>- Utilise effective contamination extraction systems</li> <li>- Good ventilation conditions</li> <li>- Minimise manual handling</li> <li>- Avoid contact with contaminated objects and instruments</li> <li>- Regularly clean the work area and equipment</li> <li>- Supervise the area to check that risk management measures are being applied</li> <li>- Train personnel for good practices</li> <li>- Keep standard personal hygiene conditions</li> </ul>																		

# Soluble Ammonium Nitrate with a Nitrogen content $\geq 34\%$ N

## Safety Data Sheet Appendices Exposure Scenario 4

<b>1</b>	<b>Title of Exposure Scenario (ES)</b>														
	End use for consumers of fertilisers, matches and fireworks														
<b>2</b>	<b>Description of activities or processes covered by the exposure scenario</b>														
	<p><b>List of all the use descriptors related to ES 3</b></p> <p>SU 21 * PC 11/12 ERC 8b/8e/10a</p> <p><b>Name/s of contributing scenario/s related to the environment and their corresponding Environmental Release Class (ERC)</b></p> <p>1. Wide dispersive indoor use of reactive substances in open systems (ERC 8b) 2. Wide dispersive outdoor use of reactive substances in open systems (ERC 8e) 3. Wide dispersive use of long-life articles and materials with low release (ERC 10a)</p> <p><b>Name(s) of contributing scenarios for the worker and their corresponding Chemical Product Category (PC)</b></p> <p>1. Explosives (PC11) 2. Fertilizers (PC 12)</p> <p>* Agency Guidance Document, Chapter R.12: Use descriptor systems: SU 21 (Uses by consumers: Private households (= general public = consumers))</p>														
<b>2.1</b>	<b>Contributing scenario (1) controlling environmental exposure for use of fertilizers, matches and fireworks by consumers (ES4)</b>														
	Environmental exposure due to the use by consumers of ammonium nitrate in fertilisers, matches and fireworks														
	Section 2.1 describes the emissions to the environment that may occur during wide dispersive indoor use of reactive substances in open systems (ERC 8b) and wide dispersive outdoor use of reactive substances in open systems (ERC 8e)														
	As this substance does fulfill criteria for classification as hazardous to the environment, the environmental risk assessment has not been carried out for this substance and therefore the conditions that affect the environment are not included during this use.														
<b>2.2</b>	<b>Contributing scenario (2) controlling consumer exposure for use of fertilizers, matches and fireworks</b>														
	All the relevant chemical product categories for this scenario identified by the PC codes in point 1 of this scenario (PC 11, PC 12) have the same operating conditions and risk management measures for personnel. Consequently they are all covered by just one contributing scenario (2).														
	<table border="1" style="width: 100%;"> <tr> <td style="width: 60%;"><b>Product characteristics</b></td> <td>Solid with low dust formation index Liquid Products containing ammonium nitrate in concentrations of <math>\geq 10\%</math> Products containing ammonium nitrate in concentrations of <math>&lt; 10\%</math></td> </tr> <tr> <td><b>Quantities used</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Frequency and duration of use or exposure</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Human factors not influenced by risk management</b></td> <td>Not relevant</td> </tr> <tr> <td><b>Other operational conditions that have an impact on worker exposure</b></td> <td>Used indoors and outdoors</td> </tr> <tr> <td><b>Conditions and measures with information and recommendations for consumer conduct</b></td> <td>Avoid splashes</td> </tr> <tr> <td><b>Conditions and measures for personal protection and hygiene</b></td> <td>1.- If concentration is <math>\geq 10\%</math>, use safety glasses 2.- If concentration is <math>&lt; 10\%</math>, no personal protection is necessary 3.- Indicate safe recommendations of use for consumers on the product labels</td> </tr> </table>	<b>Product characteristics</b>	Solid with low dust formation index Liquid Products containing ammonium nitrate in concentrations of $\geq 10\%$ Products containing ammonium nitrate in concentrations of $< 10\%$	<b>Quantities used</b>	Not relevant	<b>Frequency and duration of use or exposure</b>	Not relevant	<b>Human factors not influenced by risk management</b>	Not relevant	<b>Other operational conditions that have an impact on worker exposure</b>	Used indoors and outdoors	<b>Conditions and measures with information and recommendations for consumer conduct</b>	Avoid splashes	<b>Conditions and measures for personal protection and hygiene</b>	1.- If concentration is $\geq 10\%$ , use safety glasses 2.- If concentration is $< 10\%$ , no personal protection is necessary 3.- Indicate safe recommendations of use for consumers on the product labels
<b>Product characteristics</b>	Solid with low dust formation index Liquid Products containing ammonium nitrate in concentrations of $\geq 10\%$ Products containing ammonium nitrate in concentrations of $< 10\%$														
<b>Quantities used</b>	Not relevant														
<b>Frequency and duration of use or exposure</b>	Not relevant														
<b>Human factors not influenced by risk management</b>	Not relevant														
<b>Other operational conditions that have an impact on worker exposure</b>	Used indoors and outdoors														
<b>Conditions and measures with information and recommendations for consumer conduct</b>	Avoid splashes														
<b>Conditions and measures for personal protection and hygiene</b>	1.- If concentration is $\geq 10\%$ , use safety glasses 2.- If concentration is $< 10\%$ , no personal protection is necessary 3.- Indicate safe recommendations of use for consumers on the product labels														
<b>3</b>	<b>Estimation of exposure and reference to its source</b>														
	<p><b>Information for contributing scenario 1 (environmental exposure):</b></p> <p>No environmental assessment was completed because the substance does not fulfill criteria for classification as hazardous for the environment and therefore there is no additional assessment of environmental exposure.</p> <p><b>Information for contributing scenario 2 (exposure for personnel):</b></p> <p>A qualitative assessment has been included that concludes that this use is safe for consumers.</p> <p>The toxicological effect of this substance is eye irritation (local parameter), for which a DNEL value cannot be estimated, because there is no dose-response information available. A minimal systemic effect was only observed at doses that were so high that consumers or people would never be exposed to them (see relevant DNEL: section 8 SDS), so it was not considered necessary to make a quantitative risk assessment.</p>														
<b>4</b>	<b>Guidance for intermediate users to assess if they working within the limits set by the ES</b>														
	Additional risk management measures apart from the ones mentioned above in the contributing scenarios (2.1, 2.2) are not required to guarantee safety during this use:														
	If concentration of ammonium nitrate is $\geq 10\%$ , use safety glasses If concentration is $< 10\%$ , no personal protection is necessary														